

WITHAPERUVINS F AND G, TWO WITHANOLIDES OF *PHYSALIS PERUVIANA* ROOTS

PARTHA NEOGI, MAHENDRA SAHAI and ANIL B. RAY*

Department of Medicinal Chemistry, Institute of Medical Sciences, Banaras Hindu University, Varanasi 221005, India

(Received 23 April 1986)

Key Word Index *Physalis peruviana*; Solanaceae; withaperuvins F and G; steroidal lactones.

Abstract Withaperuvins F and G, the minor constituents of the roots of *Physalis peruviana*, have been formulated respectively, as (3R,6S,17S,20S,22R)-4 β ,5 β ,17 β ,20 α -tetrahydroxy-3 α ,6 α -oxido-1-oxoergosta-14,24-dien-22,26-olide and (17S,20S,22R)-2 β ,3 β :5 β ,6 β -diepoxy-14 α ,17 β ,20 α -trihydroxy-1-oxoergosta-24-en-22,26-olide.

INTRODUCTION

Withanolide, a term originally coined [1] to designate the steroidal lactones of *Withania somnifera*, embraces all naturally occurring C₂₈-steroidal lactones that may be chemically named as ergosta-22,26-olides or more commonly as 1-oxoergosta-2-en-22,26-olides because a great majority of this group of compounds have a steroidal Δ^2 -1-one system. Withanolides occur only in the Solanaceae and their distribution is mainly restricted to the leaves [2]. The roots of *Physalis peruviana* are, however, quite rich in withanolides and the isolation and structural determination of several new withanolides from this source were reported previously [3-6]. The present communication describes the isolation of two minor withanolides, withaperuvins F and G from the roots of *P. peruviana* and advancement of their structures as (3R,6S,17S,20S,22R)-4 β ,5 β ,17 β ,20 α -tetrahydroxy-3 α ,6 α -oxido-1-oxoergosta-14,24-dien-22,26-olide (1a) and (17S,20S,22R)-2 β ,3 β :5 β ,6 β -diepoxy-14 α ,17 β ,20 α -trihydroxy-1-oxoergosta-24-en-22,26-olide (4), respectively.

RESULTS AND DISCUSSION

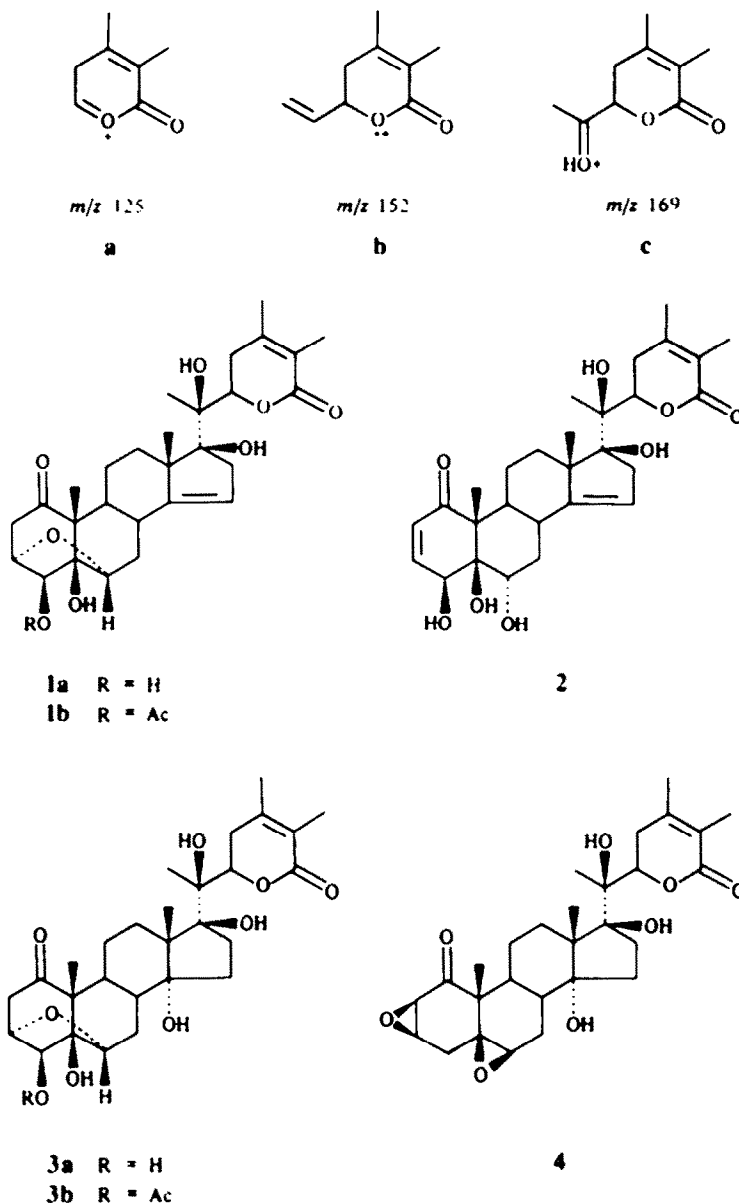
Withaperuvins F (1a), C₂₈H₃₈O₈ was recognized to be a 17,20-dihydroxywithanolide, like its congeners [3-5], from its mass spectral peaks [7, 8] at *m/z* 125 (a), 152 (b) and 169 (c) and from the diagnostic ¹H NMR signal at δ 4.66 as a double doublet [2] due to the carbinyl hydrogen at C-22. Unlike typical withanolides, withaperuvins F was found to be devoid of a Δ^2 -1-one system and its IR absorption bands at 1690 and 1700 cm⁻¹ were assigned, respectively, to α,β -unsaturated δ -lactone and cyclohexanone carbonyl groups. The ¹H NMR spectrum of withaperuvins F showed in addition to 3H singlets at δ 1.16, 1.26 and 1.28, respectively, for 18-, 19- and 21-methyl groups, two vinylic methyl singlets at δ 1.88 and 1.96 for 27- and 28-methyl groups. The lone olefinic hydrogen signal at δ 5.20 (1H, broad) discernible in the spectrum and its band width are reminiscent of the C-15 hydrogen signal of withaperuvins B (2) [4] and of physalolactone C [9]. The spectrum also showed signals for three

carbinyl hydrogens at δ 4.20, 4.33 and 4.44 in addition to one at δ 4.66 for C-22-H. Of these three signals the one at δ 4.44 was proved to be due to a secondary alcoholic group as it underwent a shift to δ 5.29 in the spectrum of its monoacetate (1b), C₃₀H₄₀O₉ (M⁺, 544), mp 132-133°, prepared by acetylation of withaperuvins F with acetic anhydride and pyridine. The remaining two carbinyl hydrogen signals at δ 4.20 and 4.33, which remained unaffected on acetylation, were considered to be due to methine hydrogens bearing an oxide bridge. The geminally coupled hydrogen signals at δ 2.43 (1H, *dd*, *J* = 18.2 and 3 Hz) and 2.60 (1H, *dd*, *J* = 18.2 and 1.6 Hz), observed in the ¹H NMR spectrum of 1b, were assigned to a ketomethylene group, the presence of which was also corroborated from a positive Zimmermann test [10]. From irradiation experiments, this ketomethylene group in 1b was proved to be a part of a COCH₂CHOCHOAcC- system as is present in withaperuvins D acetate (3b). The irradiation at signal frequency for the methine multiplet at δ 4.48 (δ 4.33 in 1a) transformed the doublet due to a hydrogen bound to the carbon bearing the acetoxy function at δ 5.29 to a sharp singlet and the ketomethylene double doublets to clean geminally coupled doublets (*J* = 18 Hz) indicating it to be a carbinyl hydrogen flanked by the ketomethylene and the carbon bearing the acetoxy function. Based on this observation and from a ¹³C NMR spectral comparison (Table I) with withaperuvins B (2) and withaperuvins D (3a) [11], structure 1a was advanced for withaperuvins F.

Rigorous evidence in favour of the proposed structure and stereochemistry of withaperuvins F came from its transformation to withaperuvins B (2) by prolonged heating with acetic acid. The identity of the enone, derived from withaperuvins F, with the naturally occurring withaperuvins B was disclosed from a comparison of their melting points, *R_f* values, IR and ¹H NMR spectra. Withaperuvins F is the second known withanolide bearing the novel 3,6-oxide bridge and the first for which the chemical evidence is provided.

Withaperuvins G (4), C₂₈H₃₈O₈ (M⁺, 502), crystallized from chloroform as needles, mp 163° and showed IR bands for hydroxyl (3350 cm⁻¹), cyclohexanone (1700 cm⁻¹), and α,β -unsaturated δ -lactone (1690 cm⁻¹) groupings. The presence of an unsaturated δ -lactone

* Author to whom correspondence should be addressed.



chromophore was also corroborated from its UV absorption maximum at 236 nm (ϵ 7400). The mass spectral peaks at m/z 125, 152 and 169 and ^1H NMR signals at δ 1.02, 1.10, 1.41, 1.86, 1.93 (3H, s each) and a one-proton triplet [12] at δ 4.85 ($J = 8$ Hz) suggested it to be a 17,20-dihydroxywithanolide. The absence of the 2-en-1-one system in withaperuv G became manifest from an examination of its ^1H NMR spectrum which did not show any olefinic hydrogen signal. Withaperuv G is not acylable and, therefore, the presence of three carbinyl hydrogen signals at δ 3.14 (br, $W/2 \approx 4.5$ Hz), 3.48 (d, $J = 4.5$ Hz), and 3.60 (overlapped dd, $J = 4.5$ and 3.9 Hz) in its ^1H NMR spectrum suggested the probable presence of a disubstituted and trisubstituted ethylene oxide system in the molecule. The presence of two epoxides in withaperuv G is also consistent with the observation that the molecule having eight oxygen atoms showed two carbonyl carbons and eight oxycarbons in its ^{13}C NMR spectrum

(Table 1). A comparison of ^1H and ^{13}C NMR parameters of the carbinyl hydrogens and oxycarbons of withaperuv G with those of withanolides of known structure and stereochemistry strongly suggested that the former is a 2 β ,3 β - and 5 β ,6 β -diepoxide like viscosalactone A (5) [12]. Accumulated evidence permitted advancement of the structure of withaperuv G as 4 which may be regarded as 2 β ,3 β -epoxide of withanolide E. In order to verify the correctness of this assumption, an attempt was made to prepare withaperuv G (4) by epoxidation of withanolide E (6a) [13] using a methanolic solution of alkaline hydrogen peroxide as has been reported by Pelletier *et al.* [12] but the product was a mixture in our case, presumably due to Michael addition of methanol. The reaction was, however, smooth and complete within 5 min and gave a near-quantitative yield of withaperuv G when epoxidation of withanolide E was performed in DMF medium.

Table 1. Carbon-13 shieldings in withaperuvins F (1a), withaperuvins G (4) and related withanolides

	(1a) (CDCl ₃ + CD ₃ OD)	(2) (C ₂ D ₅ N)	(3a) (CD ₃ OD)	(4) (CD ₃ OD)	(5) (CDCl ₃)
C-1	211.2 s	201.1	214.0	209.7	206.9
C-2	43.7 t	127.2	42.3	55.6	54.9
C-3	76.8 d	146.9	77.6	56.2	55.8
C-4	73.3 d	67.8	74.5	35.4	74.6
C-5	80.2 s	79.7	79.8	64.2	64.0
C-6	74.2 d	74.4	77.3	61.9	59.6
C-7	29.8 t	37.0	26.4	27.2	29.6
C-8	30.3 d	33.6	35.8	33.4	31.0
C-9	29.8 d	48.3	34.9	35.7	42.7
C-10	55.0 s	56.9	55.8	51.6	48.6
C-11	22.0 t	24.1	21.6	21.7	20.0
C-12	41.6 t	42.9	35.4	34.7	27.3
C-13	55.7 s	55.5	55.6	54.7	42.7
C-14	151.2 s	150.7	83.9	82.8	56.1
C-15	115.3 d	114.8	31.1	30.8	24.3
C-16	35.5 t	35.7	37.4	37.5	40.7
C-17	87.3 s	86.9	88.7	88.7	51.9
C-18	17.3 q	17.6	20.6	20.6	11.5
C-19	15.1 q	10.5	15.3	12.3	14.5
C-20	77.4 s	77.9	78.3	79.9	38.9
C-21	20.3 q	22.2	19.4	19.5	13.3
C-22	80.6 d	80.8	83.0	83.7	78.7
C-23	33.0 t	34.5	32.8	33.2	29.8
C-24	152.6 s	151.9	154.2	153.3	153.8
C-25	121.5 s	121.6	121.7	121.7	125.8
C-26	166.6 s	166.5	165.7	169.0	167.4
C-27	12.2 q	12.6	12.3	12.3	57.2
C-28	20.6 q	20.6	20.8	20.7	20.1

The orientation of the 2,3-epoxy function in semi-synthetic withaperuvins G was manifest from the observation that a nucleophile attacks the β -carbon of an enone system preferentially from the β -side in an analogous molecule. Thus when 4 β -hydroxy withanolide E (6b) was treated with methanolic alkali at room temperature for a short while, it gave a methanol adduct (7), the ¹H NMR spectrum of which showed a one-proton multiplet at δ 3.67 ($W/2 \approx 15$ Hz) due to the newly generated carbinyl hydrogen at C-3. The ¹H NMR parameters of this signal, which were found to be in perfect agreement with those reported for the C-3 hydrogen of physalactone (8) [15], indicated it to be an equatorial hydrogen in a A/B cis-fused ring system [16]. The methoxyl group at C-3 in this adduct was thus proved to be β . This observation, taken together with the fact that epoxidation of an enone with alkaline hydrogen peroxide proceeds through a mechanism analogous to the Michael addition reaction, clearly showed that the orientation of the 2,3-epoxide ring in 4 should be like the methoxy group in 7. Withaperuvins G is the second natural withanolide with a 2,3-epoxy system.

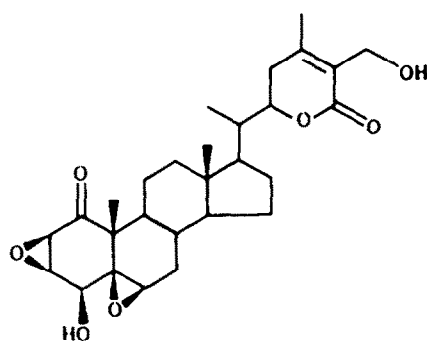
EXPERIMENTAL

Isolation procedure. Roots of *Physalis peruviana* L. were collected near Varanasi (India) during the summer of 1979. Extraction of plant material was carried out as described elsewhere [3]. Repeated chromatography (silica gel) of EtOAc

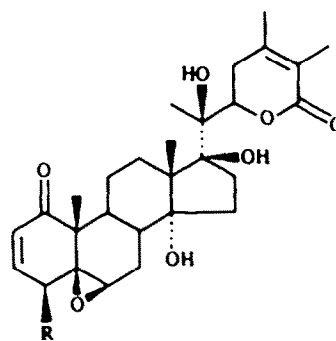
eluates of the Et₂O-soluble portion furnished a fraction which was found to be a mixture of two components on a TLC plate. Prep. TLC of this fraction on silica gel G plates yielded withaperuvins F (yield, 0.001%) and withaperuvins G (0.0005%).

Withaperuvins F (1a). Crystallized from CHCl₃ as needles, C₂₈H₃₈O₆ (M^+ 502), mp 174–175°, $[\alpha]_D^{25} + 39.70^\circ$ (c 0.26, CHCl₃); R_f 0.52 in hexane–EtOAc (1:4); UV λ_{max}^{MeOH} 228 nm (ϵ 7078); IR $\nu_{max}^{CHCl_3}$: 3450, 1710, 1700 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) δ 1.16 (3H, s, 18-CH₃), 1.26 (3H, s, 19-CH₃), 1.28 (3H, s, 21-CH₃), 1.88 (3H, s, 27-CH₃), 1.96 (3H, s, 28-CH₃), 2.32 (1H, br d, $J = 16$ Hz, 16-H), 2.91 (1H, br d, $J = 16$ Hz, 16-H), 2.49 (1H, br, 2-H), 2.62 (1H, br, 2-H), 4.20 (1H, br, 6-H), 4.33 (1H, m, 3-H), 4.44 (1H, d, $J = 6.6$ Hz, 4-H), 4.66 (1H, dd, $J = 12$ and 5 Hz, 22-H), 5.20 (1H, br, 15-H); MS m/z (rel. int. %): 502 (M^+ , 2), 359 (15), 223 (16), 186 (12), 169 (80), 152 (100), 149 (85), 125 (80), 109 (76). Found: C, 66.50; H, 7.25%. Calc. for C₂₈H₃₈O₆: C, 66.93; H, 7.57%.

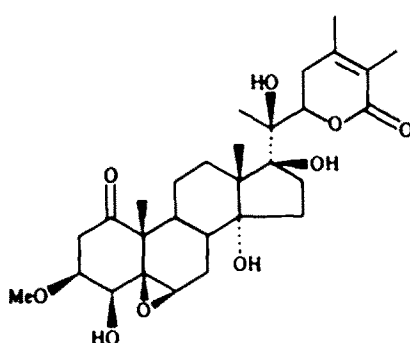
Withaperuvins F acetate (1b) was obtained (Ac₂O, py) as needles from EtOAc, mp 132–133°, $[\alpha]_D^{25} + 40.2^\circ$ (c 0.4 CHCl₃); IR $\nu_{max}^{CHCl_3}$: 3350, 1735, 1720, 1700 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) δ 1.16 (3H, s, 18-CH₃), 1.25 (3H, s, 19-CH₃), 1.29 (3H, s, 21-CH₃), 1.88 (3H, s, 27-CH₃), 1.96 (3H, s, 28-CH₃), 2.10 (3H, s, OAc), 2.29 (1H, br d, $J = 16.7$ Hz, 16-H), 2.91 (1H, br d, $J = 16.7$ Hz, 16-H), 2.34 (1H, dd, $J = 18.2$ and 3 Hz, 2-H), 2.60 (1H, dd, $J = 18.2$ and 1.6 Hz, 2-H), 4.31 (1H, dd, $J = 8.2$ and 1.2 Hz, 6-H), 4.48 (1H, m, 3-H), 5.29 (1H, d, $J = 6$ Hz, 4-H), 4.65 (1H, dd, $J = 10.6$ and 6.1 Hz, 22-H), 5.19 (1H, br s, 15-H); ¹³C NMR (CDCl₃ + CD₃OD) [C_1 – C_{28}] 209.1 (s), 44.2 (t), 76.3 (d), 75.6 (d),



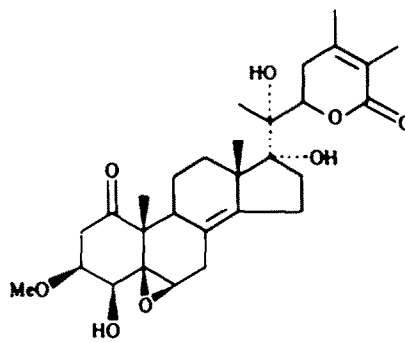
5



6a R = H
6b R = OH



7



8

78.5 (s), 71.2 (d), 29.8 (t), 30.2 (d), 29.8 (d), 54.9 (s), 22.3 (t), 41.7 (t), 56.1 (s), 150.6 (s), 115.1 (d), 35.4 (t), 87.2 (s), 17.4 (q), 15.3 (q), 77.0 (s), 20.5 (q), 80.3 (d), 33.0 (t), 152.2 (s), 121.6 (s), 165.8 (s), 12.4 (q), 20.5 (q), 172.6 (s, $-\text{COCH}_3$), 20.5 (q, $-\text{COCH}_3$). Found: C, 65.92; H, 7.29%. Calc. for $\text{C}_{30}\text{H}_{40}\text{O}_6$: C, 66.18; H, 7.35%.

Conversion of withaperuvin F (1a) to withaperuvin B (2). A solution of withaperuvin F (0.02 g) in HOAc (1 ml) was heated on a water bath for 72 hr. The solvent from the reaction mixture was removed *in vacuo* and the residue was subjected to prep. TLC. The lower band of the chromatogram, developed in C_6H_6 -EtOAc (1:4), was scraped off, eluted with EtOAc and crystallized from MeOH to yield colourless plates, mp 268–270° R_f 0.57 (C_6H_6 -EtOAc-MeOH, 2:6:1); indistinguishable from withaperuvin B (mp, co-TLC, IR); IR (KBr): 3450, 1700, 1665 cm^{-1} ; ^1H NMR (270 MHz, CDCl_3): δ 1.13 (3H, s, 21- CH_3), 1.25 (3H, s, 18- CH_3), 1.28 (3H, s, 19- CH_3), 1.88 (3H, s, 27- CH_3), 1.94 (3H, s, 28- CH_3), 2.28 (1H, br d, $J = 16$ Hz, 16-H), 2.99 (1H, br d, $J = 16$ Hz, 16-H), 4.15 (1H, dd, $J = 12$ and 4 Hz, 22-H), 4.61 (1H, t, $J = 8$ Hz, 6-H), 5.10 (1H, br s, 4-H), 5.27 (1H, br s, 15-H), 5.96 (1H, d, $J = 10$ Hz, 2-H), 6.49 (1H, d, $J = 10$ Hz, 3-H).

Withaperuvin G (4). From CHCl_3 , as needles, mp 163°; R_f 0.48 (hexane-EtOAc, 1:4); UV $\lambda_{\text{max}}^{\text{EtOH}}$ 236 nm (ϵ 7400); IR ν_{Nujol} 3350, 1700, 1690 cm^{-1} ; ^1H NMR (270 MHz, CDCl_3): δ 1.02 (3H, s, 18- CH_3), 1.10 (3H, s, 21- CH_3), 1.41 (3H, s, 19- CH_3), 1.86 (3H, s, 27- CH_3), 1.93 (3H, s, 28- CH_3), 3.14 (1H, br s, $W/2 \approx 4.5$ Hz, 6-H), 3.48 (1H, d, $J = 4.5$ Hz, 2-H), 3.60 (1H, overlapped dd, $J = 4.5$ and 3.9 Hz, 3-H), 4.85 (1H, t, $J = 8$ Hz, 22-H); MS m/z (rel. int. %): 502 (M^+ , 3), 359 (38), 315 (25), 301 (28), 223 (38), 213

(67), 169 (90), 152 (100), 125 (90), 109 (90), 105 (80), 91 (94). Found: C, 66.75; H, 7.51%. Calc. for $\text{C}_{28}\text{H}_{38}\text{O}_6$: C, 66.93; H, 7.57%.

Epoxidation of withanolide E (6a). To a solution of withanolide E (0.05 g) in HCONMe_2 (2 ml) was added H_2O_2 (30%, 2 ml) and a pinch of KOH (ca 0.01 g) and the reaction mixture was kept at room temperature for 5 min. Dilution of the reaction mixture with water (10 ml), extraction with Et_2O (3×100 ml), chromatography of the ether extract over silica gel and elution with C_6H_6 -EtOAc (2:3) yielded a solid (0.05 g) which crystallized from CHCl_3 as fine needles, mp 161–163°; IR ν_{Nujol} 3350, 1700, 1690 cm^{-1} ; ^1H NMR (270 MHz, CDCl_3): δ 1.02 (3H, s, 18- CH_3), 1.10 (3H, s, 21- CH_3), 1.42 (3H, s, 19- CH_3), 1.87 (3H, s, 27- CH_3), 1.93 (3H, s, 28- CH_3), 3.14 (1H, br s, $W/2 \approx 4.5$ Hz, 6-H), 3.48 (1H, d, $J = 4.5$ Hz, 2-H), 3.59 (1H, overlapped dd, $J = 4.5$ and 4 Hz, 3-H), 4.85 (1H, t, $J = 8$ Hz, 22-H). The compound was indistinguishable from withaperuvin G (mmp, co-TLC, IR).

Formation of methanol adduct (7) from 4 β -hydroxywithanolide E (6b). To a solution of 4 β -hydroxywithanolide E (0.1 g) in MeOH (10 ml) was added 1 N methanolic KOH (1 ml) and the reaction mixture was left at room temp. for 1 hr. The reaction mixture was then neutralized with cold dilute HCl and freed from solvent under reduced pressure to furnish a residue which was taken in water and extracted with CHCl_3 . The CHCl_3 extract was washed, dried and crystallized from MeOH to yield 7, mp 185°; ^1H NMR (270 MHz, CDCl_3): δ 1.02 (3H, s, 18- CH_3), 1.29 (3H, s, 19- CH_3), 1.42 (3H, s, 21- CH_3), 1.87 (3H, s, 27- CH_3), 1.93 (3H, s, 28- CH_3), 2.70 (1H, dd, $J = 15$ and 4.5 Hz, 2-H), 2.87 (1H, dd, $J = 15$ and 7 Hz, 2-H), 3.29 (1H, br s, $W/2 \approx 5$ Hz, 6-H), 3.33 (3H,

s, $-\text{OCH}_3$), 3.48 (1H, br s, $W/2 \approx 6.5$ Hz, 4-H), 3.67 (1H, m, $W/2 \approx 15$ Hz, 3-H), 4.85 (1H, dd, $J = 9$ and 7 Hz, 22-H). Found: C, 64.88; H, 7.78%. Calc. for $\text{C}_{29}\text{H}_{42}\text{O}_9$: C, 65.17; H, 7.86%.

Acknowledgements—We thank Dr. B. C. Das, Research Director, I.C.N.S., Gif-Sur-Yvette, France and Dr. S. C. Pakrashi, Director, Institute of Chemical Biology, Calcutta, for mass spectral analysis and to Professor S. W. Pelletier, Director, Institute for Natural Products Research, The University of Georgia, Georgia, U.S.A., for supplying ^1H NMR spectrum of viscosalactone A. P.N. thanks C.C.R.A.S., New Delhi for a fellowship. M.S. is thankful to C.S.I.R., New Delhi for financial assistance.

REFERENCES

1. Kirson, I., Glotter, E., Abraham, A. and Lavie, D. (1970) *Tetrahedron* **26**, 2209.
2. Glotter, E., Kirson, I., Lavie, D. and Abraham, A. (1978) in *Bio-organic Chemistry* (van Tamelen, ed.) Vol. 2, p. 57. Academic Press, New York.
3. Frolow, F., Ray, A. B., Sahai, M., Glotter, E., Gottlieb, H. E. and Kirson, I. (1981) *J. Chem. Soc. Perkin I* 1029.
4. Sahai, M., Neogi, P., Ray, A. B., Oshima, Y. and Hikino, H. (1981) *Heterocycles* **19**, 37.
5. Bagchi, A., Neogi, P., Sahai, M., Ray, A. B., Oshima, Y. and Hikino, H. (1984) *Phytochemistry* **23**, 853.
6. Sahai, M. and Neogi, P. (1984) *J. Indian Chem. Soc.* **61**, 171.
7. Tchesche, R., Schwang, H., Langler, H. W. and Snatzke, G. (1966) *Tetrahedron* **22**, 1129.
8. Glotter, E., Kirson, I., Abraham, A. and Lavie, D. (1973) *Tetrahedron* **29**, 1353.
9. Ali, A., Sahai, M., Ray, A. B. and Slatkin, D. J. (1984) *J. Nat. Prod.* **47**, 648.
10. Brand, J. C. D. and Scott, A. I. (1963) *Elucidation of Structure by Physical and Chemical Methods* (Bentley, K. W. ed.) *Techniques of Organic Chemistry Series* (Weissberger, A. ed.) Vol. 11, Part 1, p. 125, Wiley, New York.
11. Sahai, M., Ali, A., Ray, A. B., Slatkin, D. J. and Kirson, I. (1983) *J. Chem. Research* **S** 152.
12. Pelletier, S. W., Gebeychu, G., Nowacki, J. and Mody, N. V. (1981) *Heterocycles* **15**, 317.
13. Lavie, D., Kirson, I., Glotter, E., Rabinovich, D. and Shakked, Z. (1972) *J. Chem. Soc. Chem. Commun.* 877.
14. Gottlieb, H. E. and Kirson, I. (1981) *Org. Magn. Res.* **16**, 20.
15. Maslennikova, V. A., Tursunova, R. N. and Abubakirov, N. K. (1977) *Khim. Priir. Soedin* 531.
16. Jackman, L. M. and Sternhell, S. (1969) *Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry*, p. 288. Pergamon Press, London.